Module 3: Chemical Safety and Hazard Management

3.2

Case study on hazard management

Resource Person

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Case Study 1

Case Study: Chemical Spill in the Laboratory

Scenario:

A first-year pharmacy student is working in the college laboratory with concentrated sulfuric acid (H₂SO₄). While transferring the acid from a stock bottle into a beaker, the student accidentally knocks the beaker over, spilling sulfuric acid onto the lab bench and floor. Some acid also splashes on the student's gloves and lab coat.



Hazards Identified:

- > Chemical hazard: Corrosive nature of sulfuric acid
- > Physical hazard: Risk of slip and fall due to liquid spill
- > Health hazard: Potential chemical burns from skin exposure

Steps Taken (Hazard Management)

- > Immediate Response
- Containment and Cleanup
- Post-Incident Actions

Immediate Response

- > The student calls for help and moves away from the spill.
- > The instructor ensures the student removes contaminated gloves and rinses affected skin with water.
- > The lab is evacuated to avoid further exposure.

Containment and Cleanup

- ➤ The instructor dons proper PPE and uses a **spill kit** with acid-neutralizing agent (sodium bicarbonate).
- > The spill is safely neutralized and cleaned with absorbent material.
- Waste is collected in a labeled hazardous waste container.



Post-Incident Actions

- ➤ The student is examined by the medical officer for minor skin irritation.
- > An incident report is filed.
- ➤ A refresher training session on chemical handling and spill management is scheduled for all students.



Lessons Learned

- > Importance of using small volumes of hazardous chemicals when possible.
- > Need for proper lab technique and awareness.
- Availability and use of PPE and spill response tools can prevent injury and escalation.
- > Regular training and drills are key to building a safety-first culture.



Case Study 2

Case Study: Chemical Spill in a College Laboratory

Background

In a college chemistry lab, a student was preparing a solution using concentrated sulfuric acid. While transferring the acid from the reagent bottle to a beaker, the student accidentally knocked over the bottle, spilling acid on the workbench and floor.



Incident

- > The student was **not wearing gloves** or **a lab coat**.
- > The acid splashed on the student's hand and wrist, causing immediate burning sensation.
- > The lab had a **spill kit** and **eyewash station**, but the student was unsure how to use them.
- > The **teacher acted quickly**, using the spill kit to neutralize the acid and guided the student to the **safety shower**.
- > The student was taken to the medical room and later recovered fully.

Hazards Involved

- > Chemical hazard: Concentrated sulfuric acid (corrosive)
- > Physical hazard: Spill leading to potential slips
- > Human error: Lack of PPE and poor chemical handling

What Went Wrong

- ➤ Inadequate PPE
- > Poor chemical transfer technique
- > Lack of student training on emergency response

Corrective Measures

- Mandatory PPE policy reinforcement
- > Regular training on chemical handling and spill response
- > Clear labeling and instructions near emergency equipment
- > Implementing pre-lab safety briefings



Learning Points

- > Always wear appropriate PPE when handling hazardous chemicals.
- ➤ Be trained in **using emergency equipment** like spill kits and eyewash stations.
- > Conduct **risk assessments** and follow SOPs before starting lab work.

